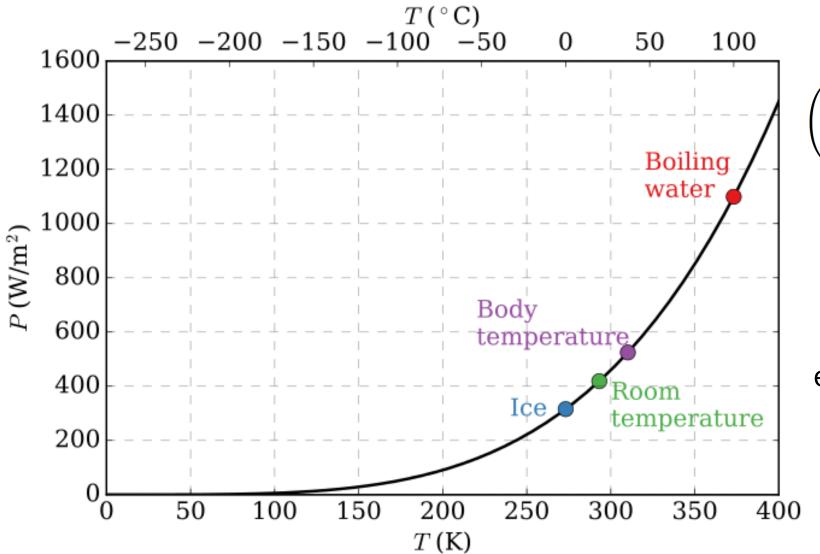
Wärmestrahlung – Stefan-Boltzmann Gesetz



Thermodynamik

$$\left(\frac{\partial U}{\partial V}\right)_T = T\left(\frac{\partial p}{\partial T}\right)_V - p$$

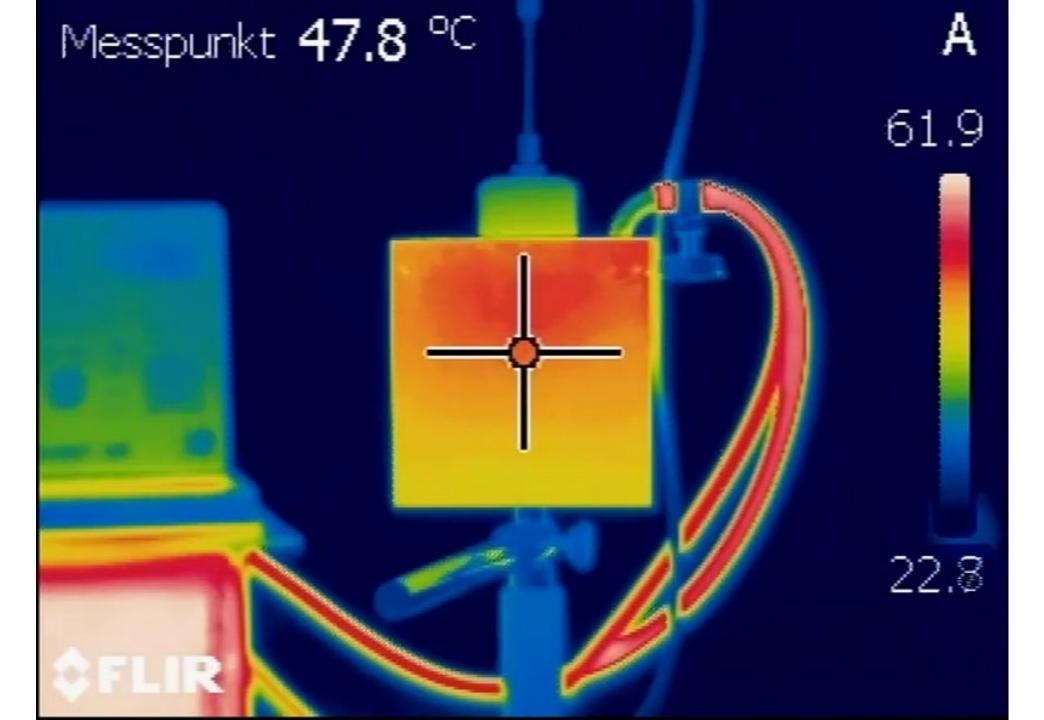
Elektrodynamik

$$p = \frac{u}{3}$$

ergibt zusammengenommen

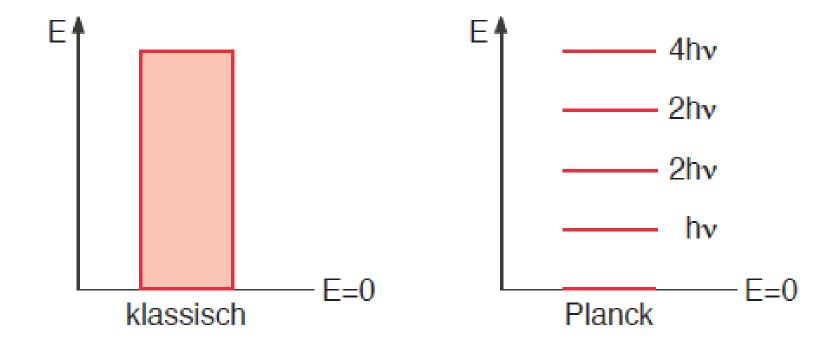
$$P = \sigma \cdot A \cdot T^4$$

Messpunkt **70.** ♣ °C 77.5 22.9 \$FLIR

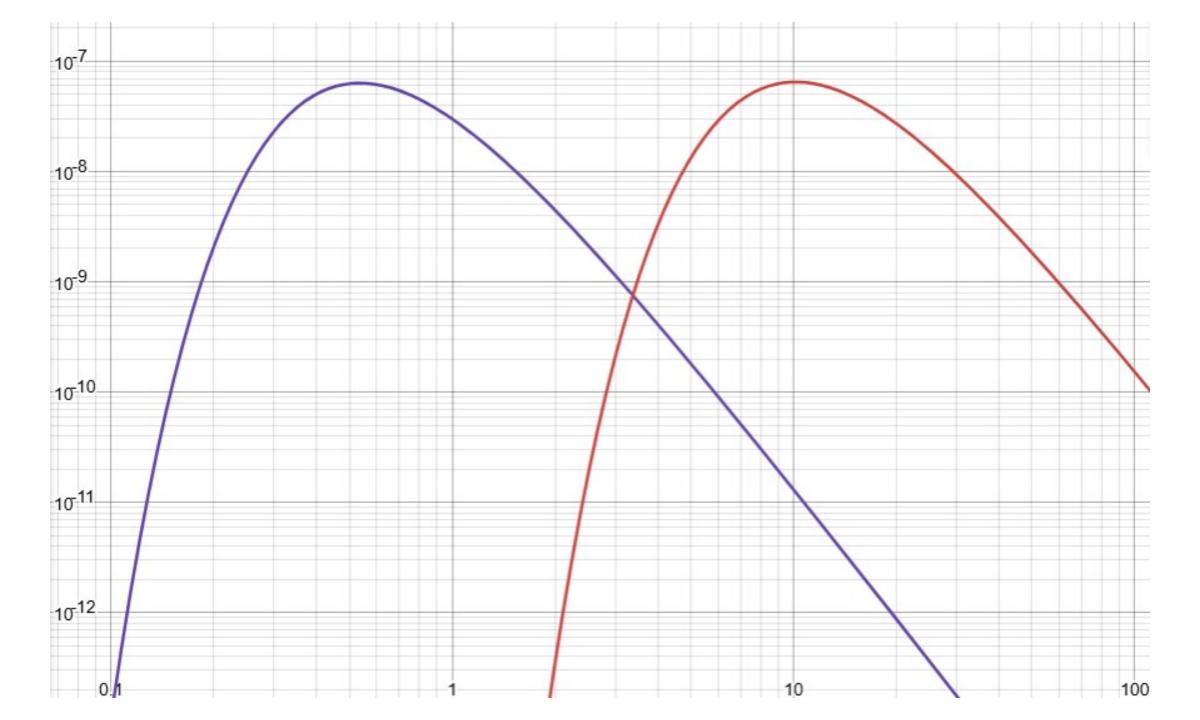


Was wenn die Strahlung nur bestimmte Energien annehmen kann?

$$E_n = n \cdot h \nu \ , n = 0, 1, 2, 3, \dots$$



Plancksche Konstante: $h = 6.626176 \times 10^{-34} \text{ Js}$



Das ergibt das Planck'sche Strahlungsgesetz für einen schwarzen Körper

-

$$u(\nu, T)d\nu = (8\pi h\nu^3)/c^3 \left[\exp(h\nu/k_B T) - 1\right]^{-1} d\nu$$
$$u(\lambda, T)d\lambda = (8\pi hc)/\lambda^5 \left[\exp(hc/\lambda k_B T) - 1\right]^{-1} d\lambda$$

Plancksche Konstante: $h = 6.626176 \times 10^{-34} \text{ Js}$

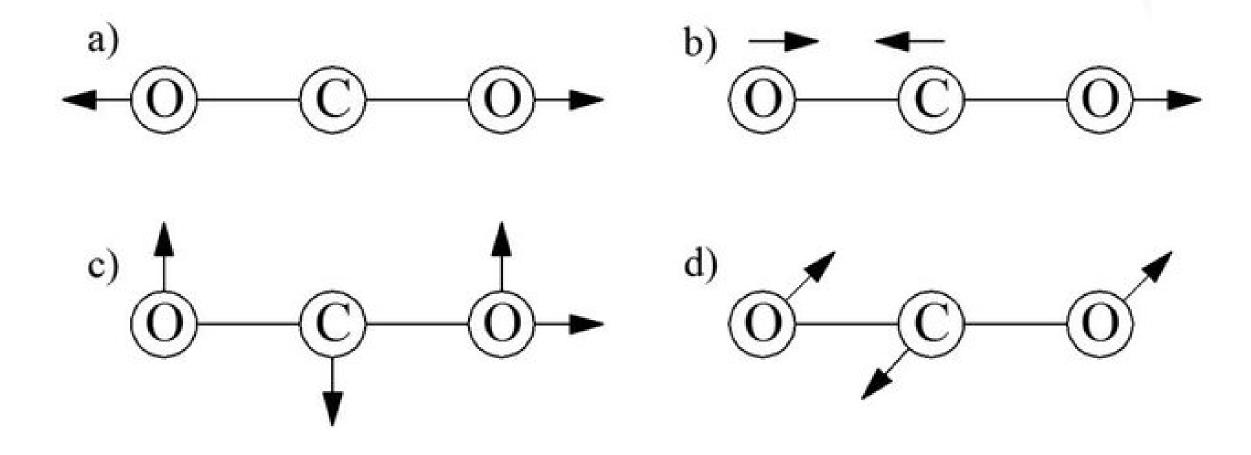
Und daraus folgt auch das Stefan-Boltzmann'sche Gesetz

Damit können wir die Stefan-Boltzmann Konstante direkt aus dem Planck'schen Gesetz bestimmen

$$\sigma = \frac{2\pi^5 k_B^4}{15c^2 h^3} = 5.67051(19) \cdot 10^{-8} \frac{W}{m^2 K^4}$$

Wir können also aus den fundamentalen konstanten h, k und c, sowie der Solarkonstanten eine mittlere Gleichgewichtstemperatur der Erde ausrechnen (inkl Albedo), was 253 K ergäbe.

Mit den Bindungseigenschaften erhalten wir die Absorptionseigenschaften

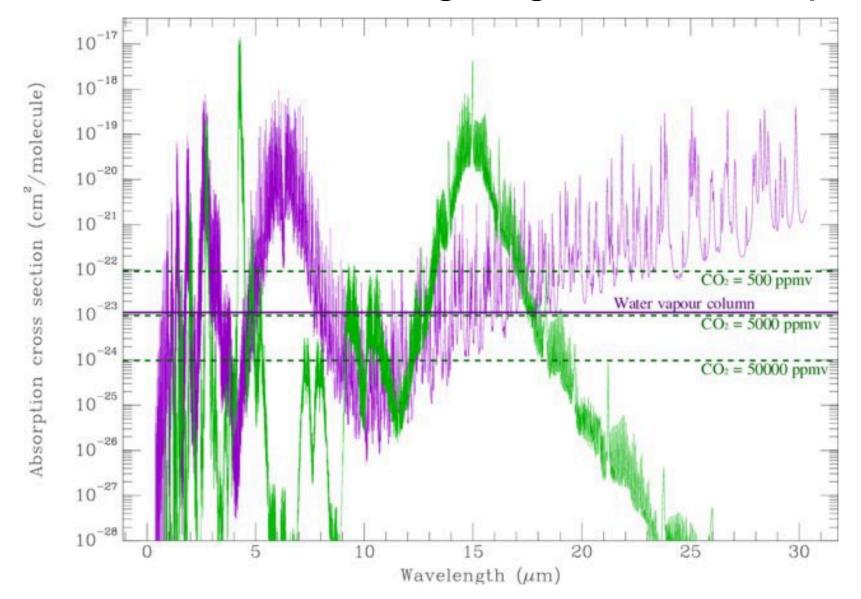


Mit den Bindungseigenschaften erhalten wir die Absorptionseigenschaften

7 110 0 0 1 p 110 1 1 0									
*Bond Lengt	ths, <i>pm</i>			[*Bond Energ	ies, <i>D</i> , (kJ/mol)			
Multiple	Bonds				Multiple	Bonds	2507. 1550	(4)5630	
C=C C≡C	134 120	O=O	121		C=C C≡C	611 837	O=O	498	
C=N	127	N=N	122		C=N	615	N=N	418	
C≡N C=O	115 121	N≡N	113		C≡N C=O	891 799	N≡N	946	
C≡O	113	S=O	133		C≡O	1072	S=0	523	
	$\lambda =$	$\frac{2\pi c}{\omega} =$	$\sqrt{\frac{4\pi^2c^2n}{k}}$	$\frac{\overline{n}}{n} = \sqrt{n}$	$\frac{4\pi^2c^2}{2}$	$\frac{2x^2m}{E}$	$=2\pi a$	$c\sqrt{rac{mc^2}{2E}}$	
$\sqrt{10}$									

 $\lambda = 20\sqrt{\frac{12}{16}}10^{-6}m \simeq 15\mu m$

Mittlere freie Weglänge und Absorptionsquerschnitt

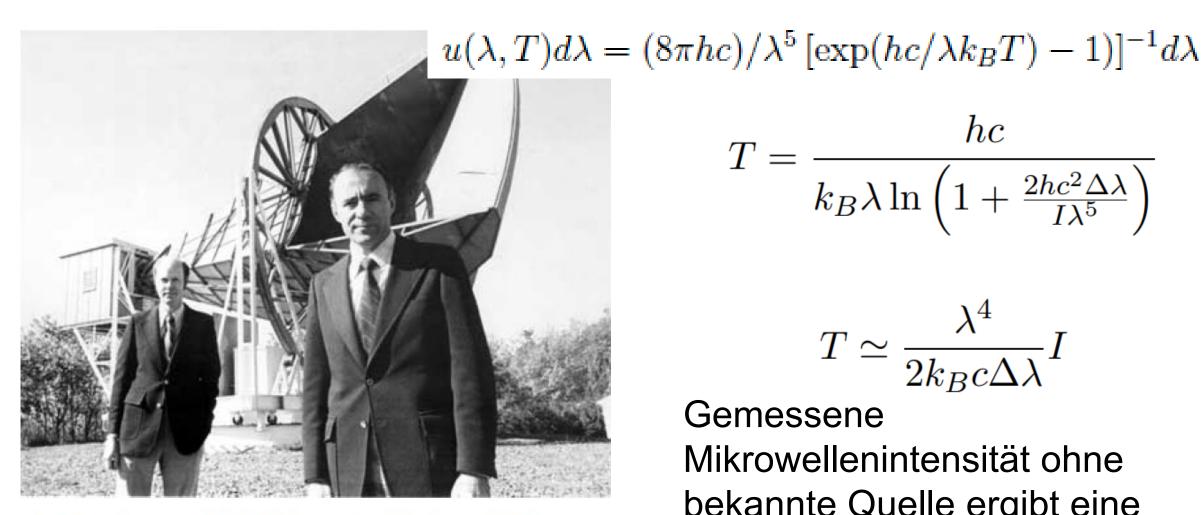


$$\ell = (\sigma n)^{-1}$$

$$dI = -In\sigma dx$$

Die Absorptionslänge in der Atmosphäre ist also etwa 30 m

Temperaturen aus Intensitäten (kosmische Hintergrundstrahlung, Wärmebildkamera)



A. Penzias und R. Wilson, Bell Labs, 1965 Nobelpreis 1978

$$T = \frac{hc}{k_B \lambda \ln \left(1 + \frac{2hc^2 \Delta \lambda}{I\lambda^5}\right)}$$

$$T \simeq \frac{\lambda^4}{2k_B c \Delta \lambda} I$$

Gemessene Mikrowellenintensität ohne bekannte Quelle ergibt eine Temperatur von 3.5±1.0 K

$$\sigma = \frac{\mu^2 E_0^2 \delta \omega}{2\hbar \omega I} = 4\pi \alpha q^2 d^2 \frac{\delta \omega}{\omega}$$

$$\sigma = \frac{4\pi}{137} \cdot 0.2^2 \cdot 0.5^2 \cdot 10^{-20} \cdot m^2 \cdot \frac{1}{3} = 2 \cdot 10^{-24} \cdot m^2$$

$$\ell = \frac{1}{n\sigma} = \frac{k_B T}{\rho p_0 \sigma} = \frac{4 \cdot 10^{-21} J}{4 \cdot 10^{-4} 10^5 Pa2 \cdot 10^{-24} m^2} = 50m$$

Wie nimmt der Druck mit der Höhe ab?

$$\frac{dp}{dh} = -g\frac{\varrho_0}{p_0}p \qquad p(h) = p_0 e^{-g\varrho_0 h/p_0}$$

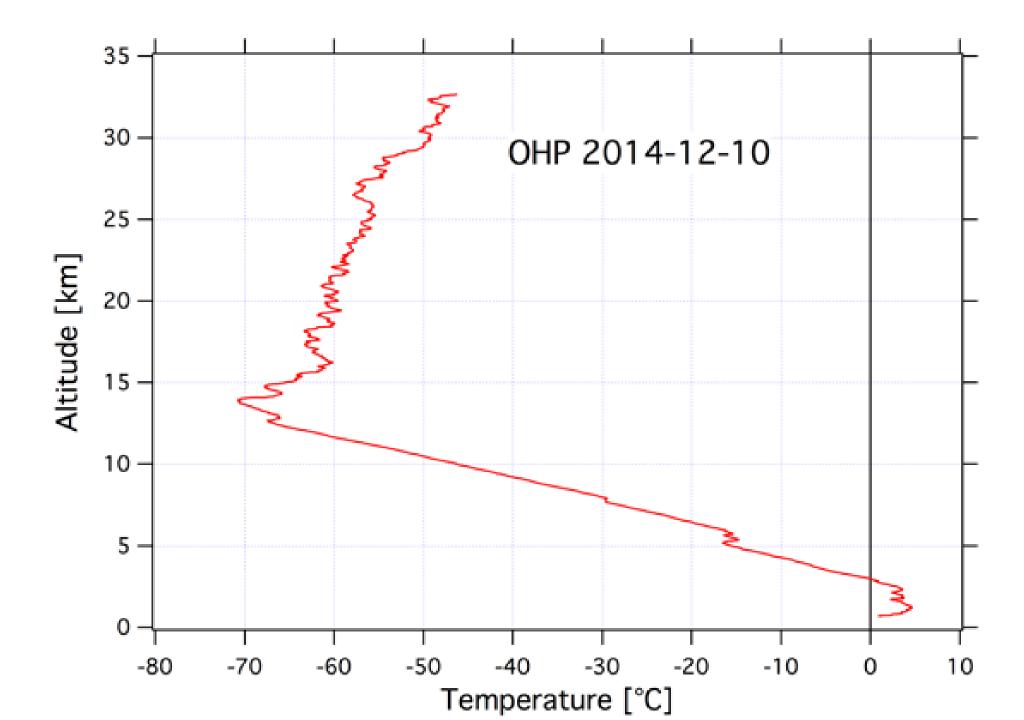
$$n = n_0 e^{-mgh/(kT)}$$
Boltzmann Verteilung!

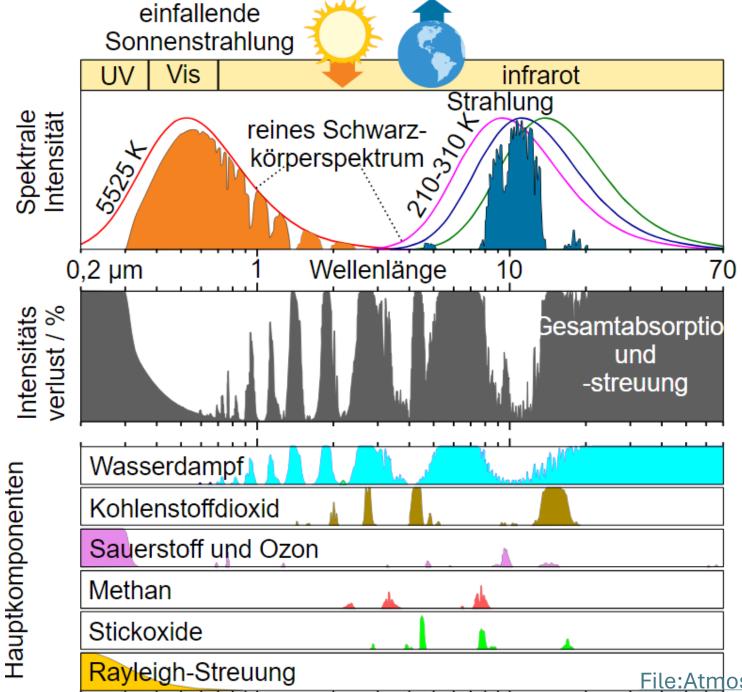
Dadurch nimmt die Absorptionslänge mit der Höhe zu!

Was macht die Temperatur in der Atmosphäre?

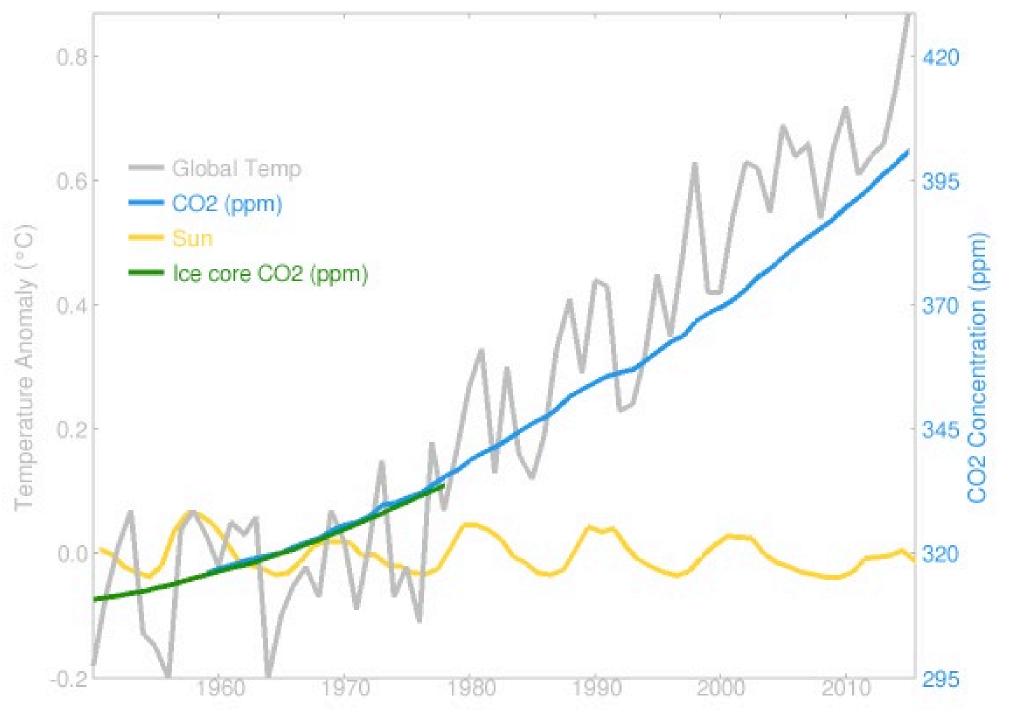
$$P\mathrm{d}V = -rac{V\mathrm{d}P}{\gamma} \hspace{1cm}
ho c_\mathrm{p} \mathrm{d}T - \mathrm{d}P = 0 \ mc_\mathrm{v} \mathrm{d}T - rac{V\mathrm{d}P}{\gamma} = 0 \hspace{1cm} \mathrm{d}P = -
ho g\mathrm{d}z$$

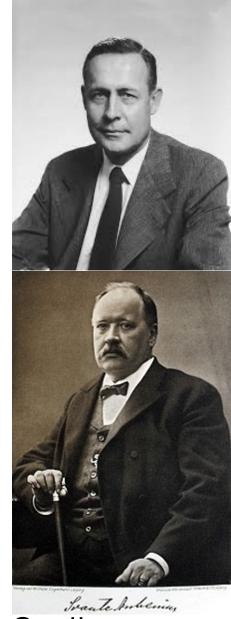
$$-rac{\mathrm{d}T}{\mathrm{d}z}=rac{g}{c_{\mathrm{p}}}$$



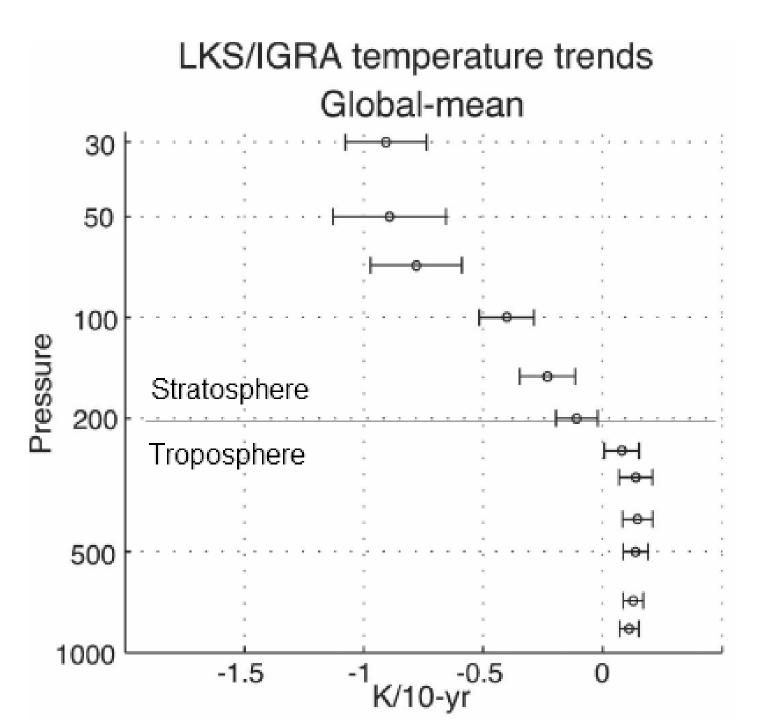


<u>File:Atmospheric Transmission.svg - Wikimedia Commons</u>



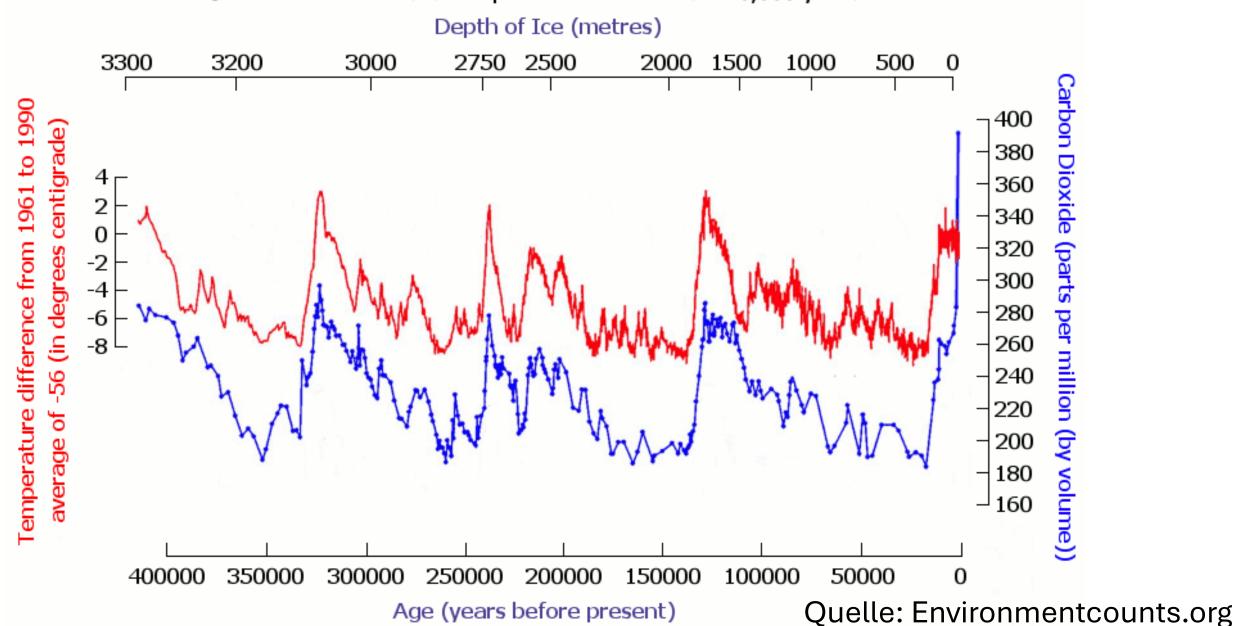


Quelle: Realclimate.org



The Vostok (Antarctica) Ice Core Record.

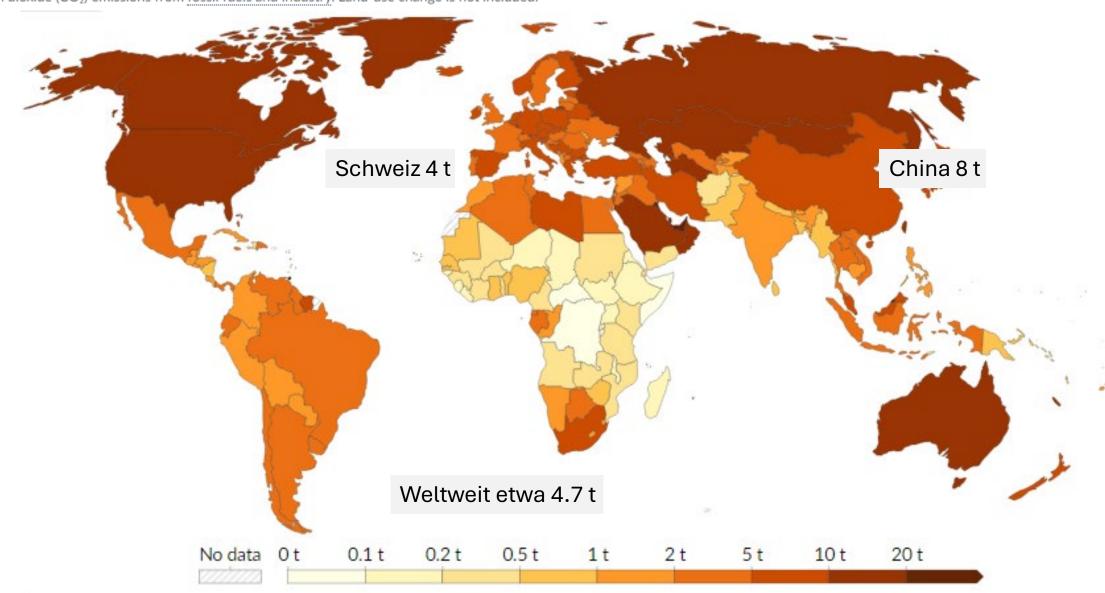
Carbon Dioxide versus Temperature for the last 420,000 years.



Per capita CO₂ emissions, 2022

Our World in Data

Carbon dioxide (CO₂) emissions from fossil fuels and industry. Land-use change is not included.



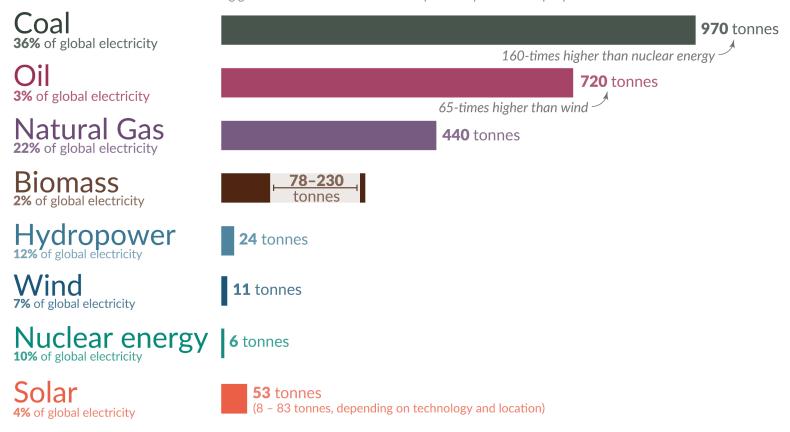
What are the





Greenhouse gas emissions

Measured in emissions of CO₂-equivalents per gigawatt-hour of electricity over the lifecycle of the power plant. 1 gigawatt-hour is the annual electricity consumption of 150 people in the EU.



Death rates from fossil fuels and biomass are based on state-of-the art plants with pollution controls in Europe, and are based on older models of the impacts of air pollution on health. This means these death rates are likely to be very conservative. For further discussion, see our article: OurWorldinData.org/safest-sources-of-energy. Electricity shares are given for 2021. Data sources: Markandya & Wilkinson (2007); UNSCEAR (2008; 2018); Sovacool et al. (2016); IPCC AR5 (2014); UNECE (2022); Ember Energy (2021).

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What are the safest and cleanest sources of energy?

Greenhouse gas emissions

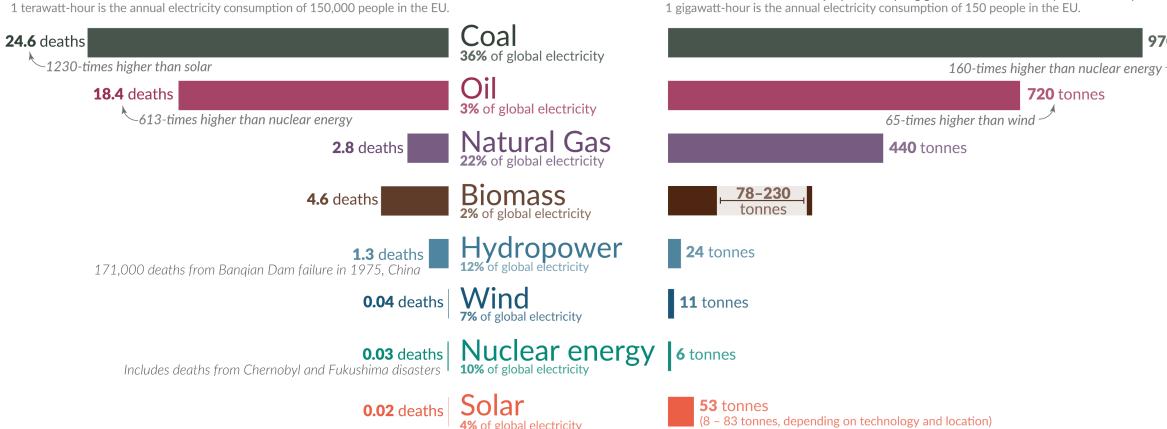
Measured in emissions of CO,-equivalents per gigawatt-hour of electricity over the lifecycle of the power plant.



970 tonnes

Death rate from accidents and air pollution

Measured as deaths per terawatt-hour of electricity production. 1 terawatt-hour is the annual electricity consumption of 150,000 people in the EU.



Death rates from fossil fuels and biomass are based on state-of-the art plants with pollution controls in Europe, and are based on older models of the impacts of air pollution on health. This means these death rates are likely to be very conservative. For further discussion, see our article: OurWorldinData.org/safest-sources-of-energy. Electricity shares are given for 2021. Data sources: Markandya & Wilkinson (2007); UNSCEAR (2008; 2018); Sovacool et al. (2016); IPCC AR5 (2014); UNECE (2022); Ember Energy (2021).

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720 tonnes